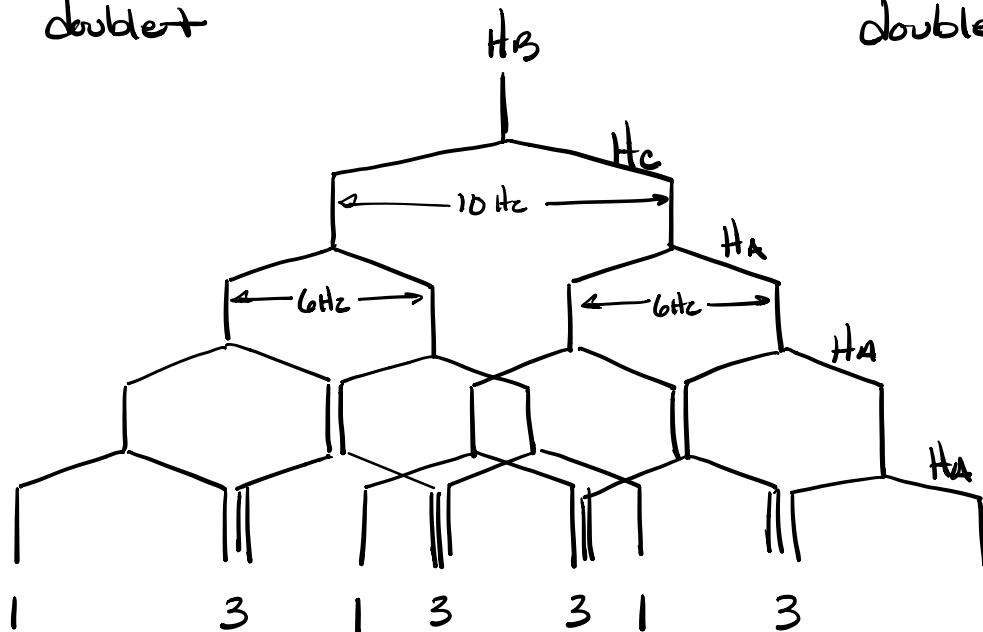
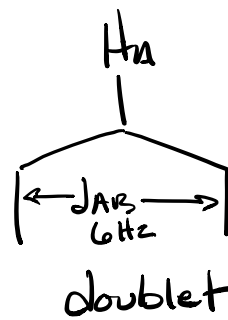
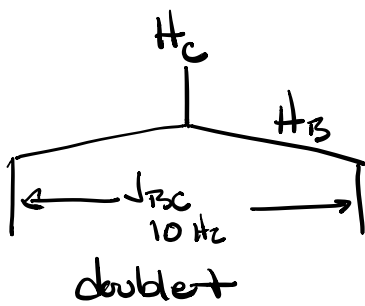
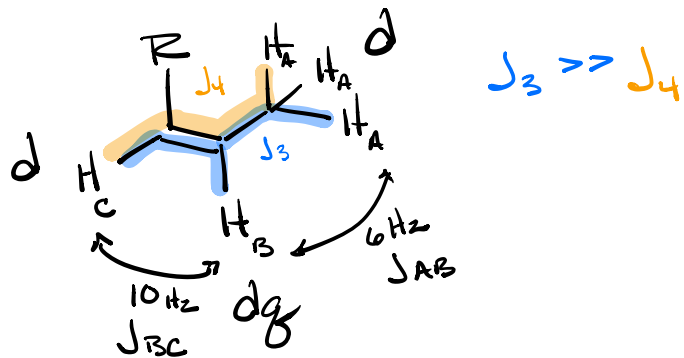
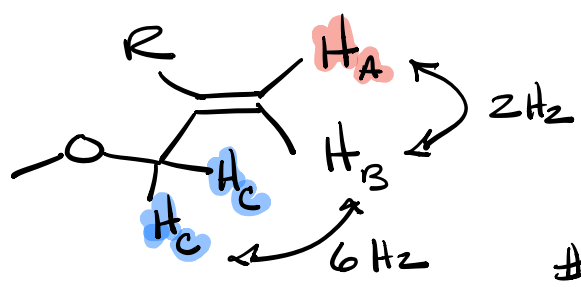
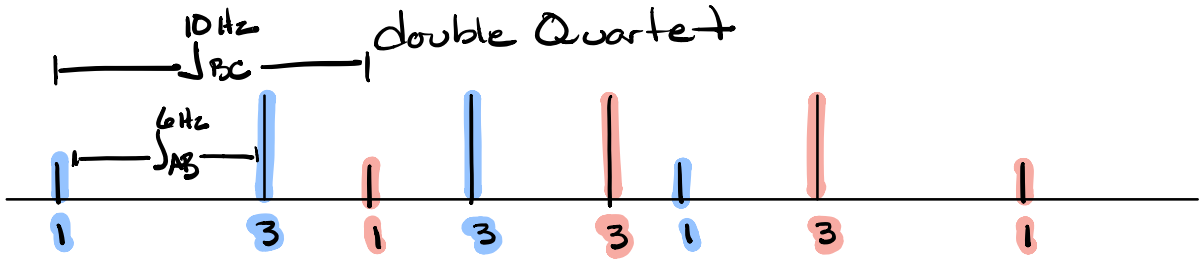
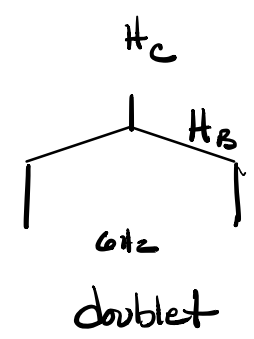
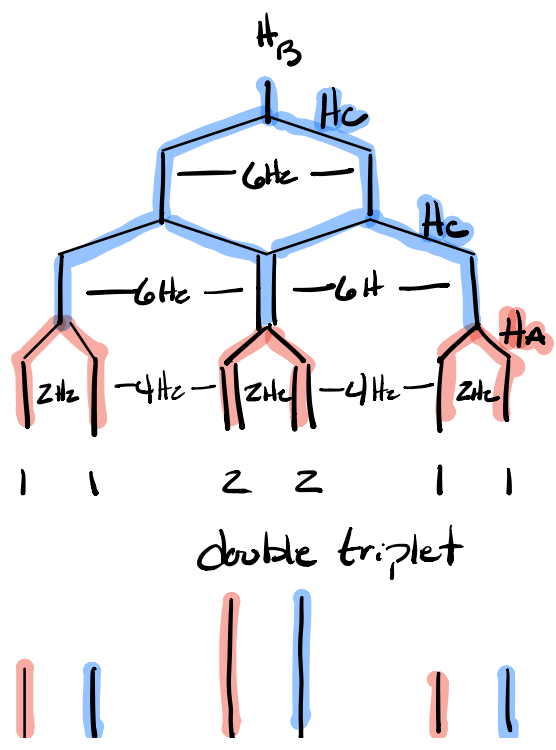
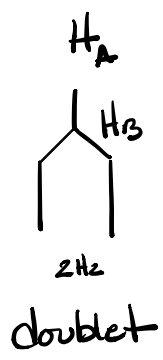


Spin-Spin Coupling





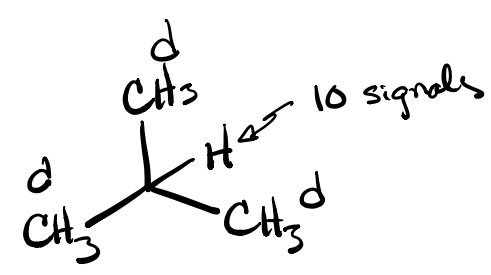
of lines in diagram = intensity of signal



Pascal's Triangle

Where all J-values are equal in the tree diagram

		1						
		1		1				doublet
		1	2	1				triplet
		1	3	3	1			quartet
		1	4	6	4	1		pentet
		1	5	10	10	5	1	sextet



#2 on Webspectra

Keys to problem solving.

- Units of unsaturation

$C_nH_{2n+2+N-x}$ formula for saturation




$2(8) + 2 = 18$ hydrogens if saturated

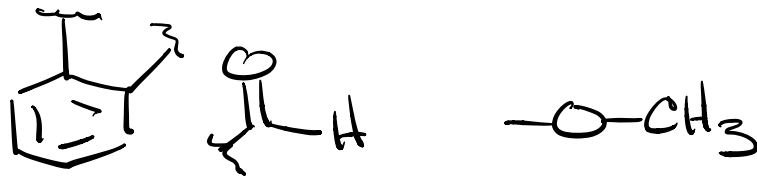
- 8 hydrogens in molecule

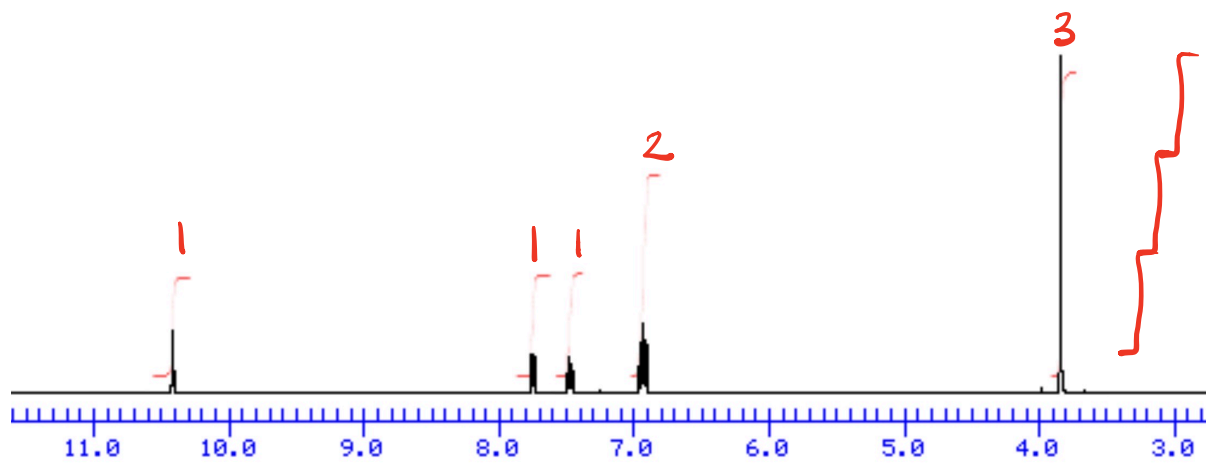
$2 \overline{) 10}$ hydrogen missing
5 units



Data Table

<u>ppm</u>	<u>Int</u>	<u>Splitting</u>	<u># of Neighbors</u>	<u>Assignment</u>
3.85	3	s	0	EWG-CH ₃ -O-CH ₃
6.93	2	mult	?	Ar
7.48	1	t	2	Ar
7.75	1	d	1	Ar
10.42	1	s	0	



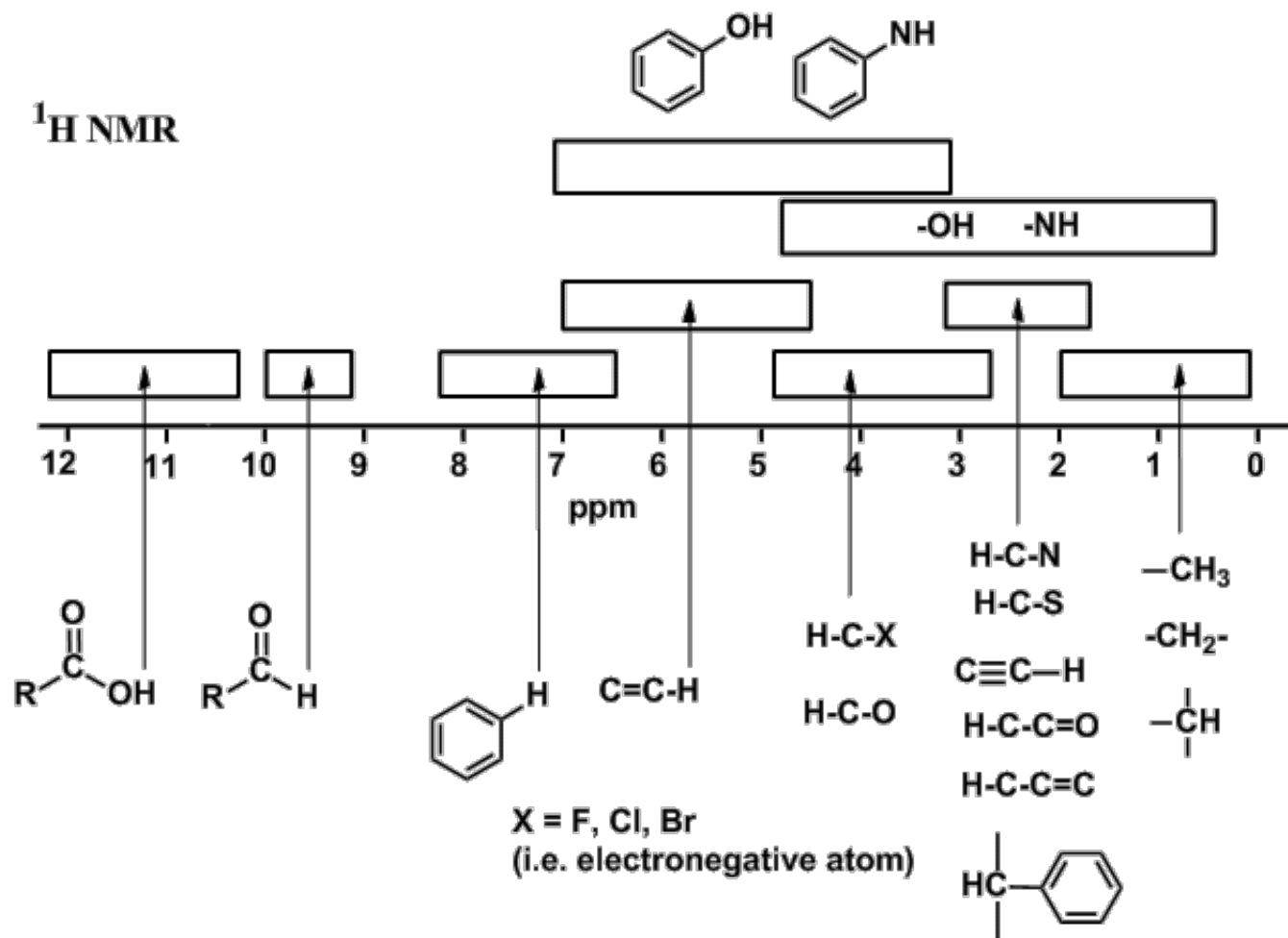


to ppm

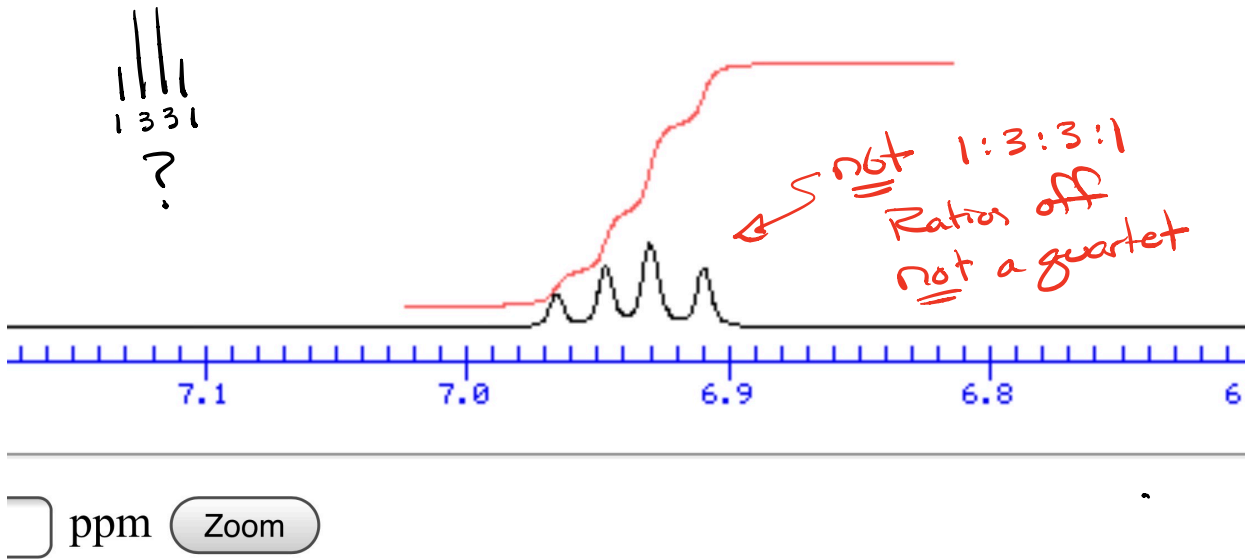
Spectrum may be 1

[Back to WebSpectra Home Page](#)

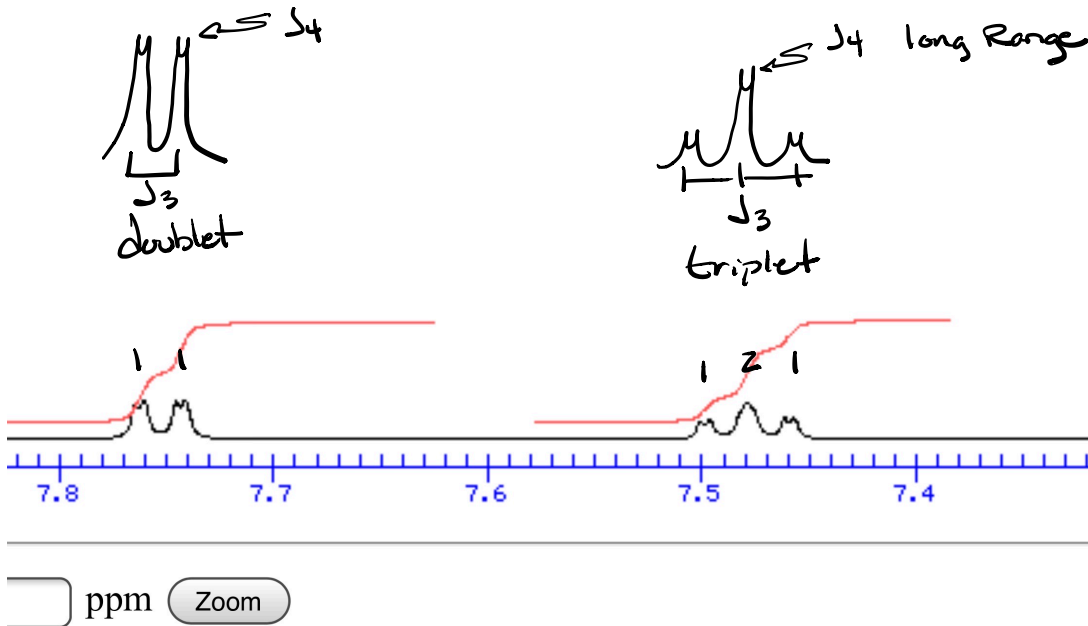
¹H NMR



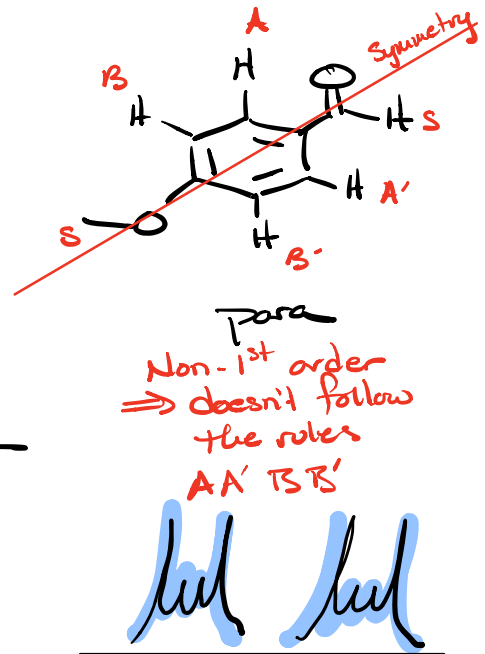
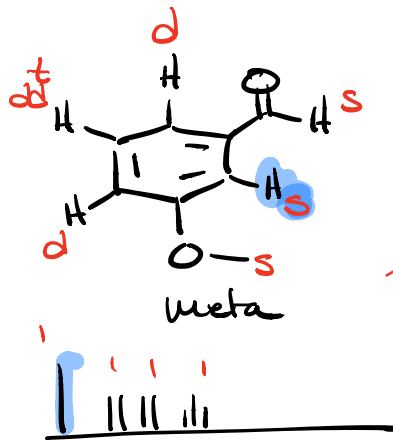
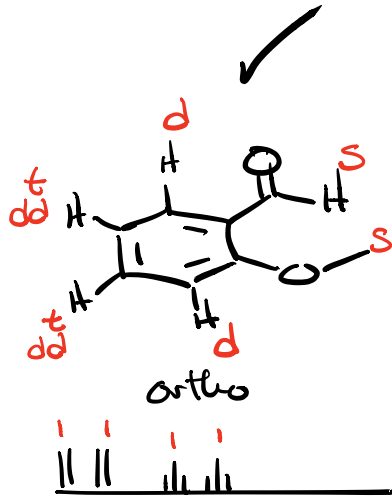
2 H's same or similar environment



[Back to WebSpectra Home Page](#)



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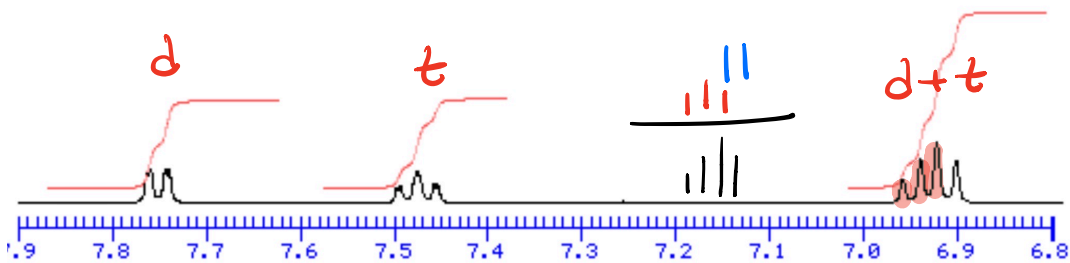


4:08 PM Thu Mar 18

webspectra.chem.ucla.edu

AA' BB' 1%

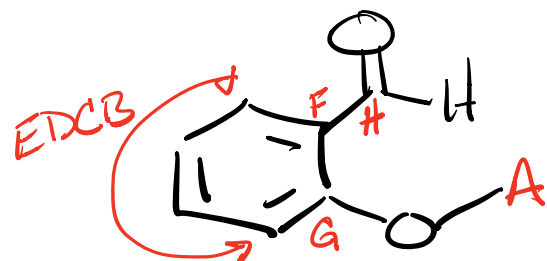
No Singlet!



ppm Zoom

Zoom Out

[Back to WebSpectra Home Page](#)



4:16 PM Thu Mar 18

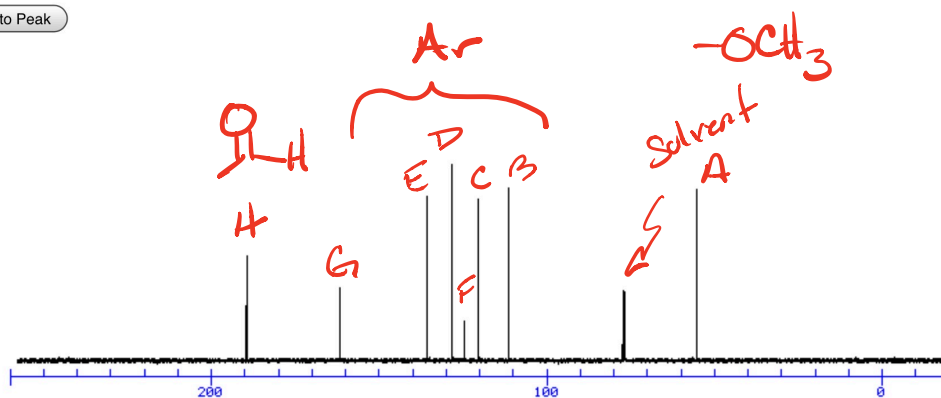
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74%

R Spectrum - $C_8H_8O_2$

[\$^1H\$ NMR Spectrum](#)
[Back to Problem](#)

8 ppm



Zoom to range: to ppm

Spectrum may be magnified 16X by clicking on peaks of in

[Back to WebSpectra Home Page](#)

2H's

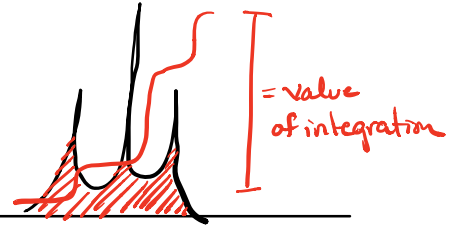


$$\frac{32,796.41}{32,796.41}$$

$$= 1$$

$$\times 2 = 2$$

3H's



$$\frac{46230.72}{32796.41}$$

$$= \sim 1.5$$

$$\times 2 = 3$$